

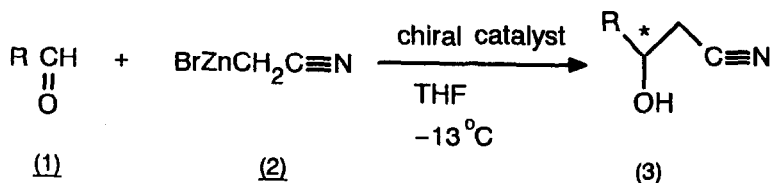
Highly Enantioselective Synthesis of β -Hydroxy Nitriles by the Cyanomethylation of Aldehydes using DPMPM as a Chiral Catalyst or Ligand

Kenso Soai,* Yuji Hirose, and Shuichi Sakata
Department of Applied Chemistry, Faculty of Science,
Science University of Tokyo, Shinjuku, Tokyo, 162 Japan

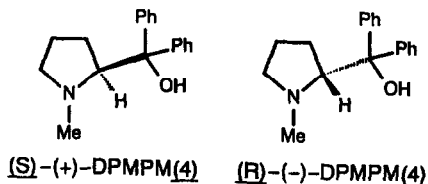
(Received 19 March 1992)

Abstract: Optically active β -hydroxy nitriles in up to 93% e.e. were obtained by the enantioselective addition of cyanomethylzinc bromide to aldehydes using DPMPM as a chiral catalyst or ligand.

Optically active β -hydroxy nitriles (3) are important compounds because cyano group can be converted into amino or carbonyl compounds.¹ However, there have been only a few reports on the preparation of optically active 3 by chemical² and biochemical³ methods. We previously reported the first enantioselective addition of cyanomethyl lithium to benzaldehyde using an aminoalcohol as chiral ligand.² However, a stoichiometric amount of the chiral ligand was required and the e.e. of the obtained β -hydroxy nitrile was moderate (40% e.e.).



- a; R=Ph
- b; R=4-MeO-C₆H₄
- c; R=2-Naphthyl
- d; R=PhCH₂CH₂
- e; R=(E)-PhCH=CH



Although the enantioselective addition of simple dialkylzincs to aldehydes using chiral catalysts are reported,⁴ enantioselective addition of organozinc reagents with functional groups other than alkyls is a challenging problem.⁵

We report here a highly enantioselective cyanomethylation of aldehydes (1) with cyanomethylzinc bromide (2)⁶ using diphenyl(1-methyl-2-pyrrolidinyl)methanol (DPMPM)(4)⁷ as a chiral catalyst or ligand.

When benzaldehyde (1a) was reacted with cyanomethylzinc bromide (2) in the presence of 1 mol equiv. of (S)-(+)-4 in THF at -13 °C, (S)-(-)-3-hydroxy-3-phenylpropionitrile (3a) with 93% e.e. was obtained in 76% yield (Table, entry 1). E.e. was determined by HPLC analysis using a chiral column. The result shows that the reagent 2 attacked from the Si face of the aldehyde (1). This stereochemical course is the same with that of dialkylzincs.⁷ On the other hand, in the presence of (R)-(-)-4, (R)-(-)-3a with 93% e.e. was obtained (entry 2). Other aryl aldehydes (1b and 1c) were also cyanomethylated in 87-88% e.e.'s (entries 5 and 6). The reaction with 3-phenylpropionaldehyde (1d) (phenyl substituted aliphatic aldehyde) afforded (R)-(+)-3d with 74% e.e. (entry 7). The reaction with cinnamaldehyde (1e) (α, β -unsaturated aldehyde) afforded (S)-(-)-3 (1,2-addition product) with 78% e.e. (entry 8).

Increasing interest has been centered on catalytic asymmetric synthesis.⁸ Because cyanomethylzinc bromide adds to aldehydes without catalyst,⁶ it was a question whether a catalytic amount of DPMPM (4) affords optically active 3. It was found that even the catalytic amount (0.3 mol equiv.) of (S)-(+)-4 afforded optically active (S)-(-)-3a with 78% e.e. (entry 3). Thus, the role of 4 is not only a chiral ligand but a chiral catalyst.

A typical procedure is as follows: (i) Preparation of cyanomethylzinc bromide (2)⁶: A THF solution (2 ml) of bromoacetonitrile (0.480g, 4 mmol) was added to a Zn-Cu couple (Zn content 91%), and the mixture was stirred for 1 h at room temperature. (ii) Enantioselective cyanomethylation: A THF solution (1 ml) of (S)-(+)-4 was added to 2 and the mixture was stirred for 1 h. The mixture was cooled to -13 °C in ethylene glycol bath and benzaldehyde (1a) (0.106g, 1 mmol) was added. The reaction mixture was stirred at -13 °C for 16 h, and was quenched with 1 M HCl. The mixture was extracted with ethyl acetate, and the extract was dried over anhydrous Na₂SO₄, and the solvent was evaporated under a reduced pressure. The residue was purified by silica gel TLC [developing solvent: hexane-AcOEt 2:1 (v/v) then di-

Table. Enantioselective Synthesis of β -Hydroxy Nitriles (3) by the Cyanomethylation of Aldehydes (1) using DPMPM (4) as a Chiral Catalyst or Ligand.

| Entry ^a | RCHO(1) | (4)(Molar ratio 4/1) | Time (h) | 3 | | Config. |
|--------------------|---------|-----------------------|----------|----------|----------------------|----------|
| | | | | Yield(%) | E.e.(%) ^b | |
| 1 | 1a | (<u>S</u>)-(+)(1.0) | 16 | 76 | 93 | <u>S</u> |
| 2 | 1a | (<u>R</u>)-(-)(1.0) | 16 | 77 | 93 | <u>R</u> |
| 3 ^c | 1a | (<u>S</u>)-(+)(0.3) | 17 | 45 | 78 | <u>S</u> |
| 4 ^d | 1a | (<u>S</u>)-(+)(0.3) | 16 | 61 | 75 | <u>S</u> |
| 5 | 1b | (<u>S</u>)-(+)(1.0) | 17 | 70 | 88 | |
| 6 | 1c | (<u>S</u>)-(+)(1.0) | 15 | 82 | 87 | |
| 7 | 1d | (<u>S</u>)-(+)(1.0) | 16 | 58 | 74 | <u>R</u> |
| 8 | 1e | (<u>S</u>)-(+)(1.0) | 16 | 59 | 78 | <u>S</u> |

^a Unless otherwise noted, molar ratio 2/1 = 4.0.

^b Determined by HPLC analyses using a chiral column (Chiralcel OF).

^c Molar ratio 2/1 = 2.4.

^d Molar ratio 2/1 = 3.0.

chloromethane]. (S)-(-)-3a was obtained in 76%.

As described above, highly enantioselective cyanomethylation of aldehydes was achieved using 4 as a chiral catalyst or ligand. Optically active β -hydroxy nitriles (3) are obtained in high e.e.'s. Either enantiomer of 3 of the desired configuration can be synthesized using the appropriate enantiomer of chiral 4.⁹

References

1. G. Tennant, *Comprehensive Organic Chemistry*, D. H. R. Barton, D. Ollis, Eds.; Pergamon Press: 1979; Vol. 2, p. 385.
2. K. Soai and T. Mukaiyama, *Bull. Chem. Soc. Jpn.*, 1979, 52, 3371.
3. T. Itoh, Y. Takagi, and S. Nishiyama, *J. Org. Chem.*, 1991, 56, 1521; T. Itoh, T. Fukuda, and T. Fujisawa, *Bull. Chem. Soc. Jpn.*, 1989, 62, 3851.
4. Reviews: R. Noyori and M. Kitamura, *Angew. Chem. Int. Ed. Engl.*,

- 1991, 30, 49; K. Soai and S. Niwa, *Chem. Rev.*, in press.
5. Alkynylzinc: S. Niwa and K. Soai, *J. Chem. Soc., Perkin Trans. 1*, 1990, 937; G. M. R. Tombo, E. Didier, and B. Loubinoux, *Synlett*, 1990, 547. Vinyl- and alkenyl-zinc: W. Oppolzer and R. N. Radinov, *Tetrahedron Lett.*, 1988, 29, 5645; *idem.*, *ibid.*, 1991, 32, 5777. Furylzinc: K. Soai and Y. Kawase, *J. Chem. Soc., Perkin Trans. 1*, 1990, 3214. Phenylzinc: K. Soai, Y. Kawase, and A. Oshio, *ibid.*, 1991, 1613; J. Hübscher and R. Barner, *Helv. Chim. Acta*, 1990, 73, 1068. Reformatsky reagent: K. Soai and Y. Kawase, *Tetrahedron: Asymmetry*, 1991, 2, 781; M. Guette, J. Capillon, and J.-P. Guette, *Tetrahedron*, 1973, 29, 3659.
 6. N. Goasdoue and M. Gaudemar, *C. R. Acad. Sc. Paris, Serie C*, 1969, 269, 861.
 7. K. Soai, A. Ookawa, K. Ogawa, and T. Kaba, *J. Chem. Soc., Chem. Commun.*, 1987, 467; K. Soai, A. Ookawa, T. Kaba, and K. Ogawa, *J. Am. Chem. Soc.*, 1987, 109, 7111.
 8. B. Bosnich, 'Asymmetric Catalysis,' Martinus Nijhoff, Dordrecht, 1986; 'Asymmetric Synthesis,' Ed. J. D. Morrison, Academic, Orlando, 1985, vol. 5.
 9. Compound (4) is commercially available in either enantiomeric form from Tokyo Kasei Inc.